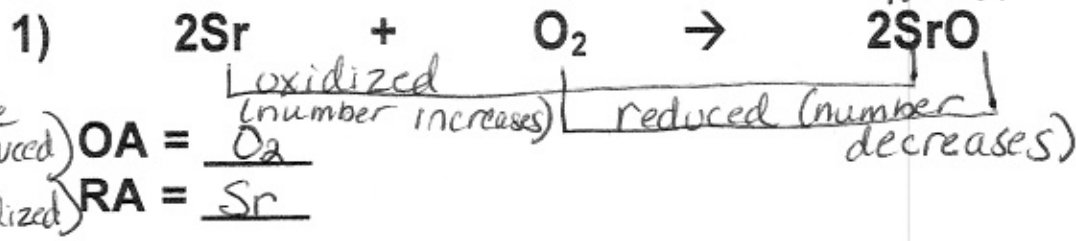


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Determine the oxidation states within each of the following compounds. Identify (using brackets) where oxidation and reduction take place, as well as which species is the reducing agent (RA) and the oxidizing agent (OA).

Uncombined elements are 0.



Loss
electrons
Oxidation

OA = _____
RA = _____



OA = _____
RA = _____



Gain
electrons
Reduction



OA = _____
RA = _____



OA = _____
RA = _____



OA = _____
RA = _____